

separated from wax by means of oil of turpentine, which dissolves the wax, while the earthy matters form a residue.

Oil of turpentine also completely separates wax from starchy substances, which, like earthy matters, do not dissolve, but form a residue. A simpler method consists in heating the wax with boiling water; the gelatinous consistency assumed by the water, and the blue coloration in presence of iodine, indicate that the wax contains starchy substances. Adulteration by means of starch and fecula is quite frequent. These substances are sometimes added to the wax in a proportion of nearly 60 per cent. To separate either, the suspected product is treated hot with very dilute sulphuric acid (2 parts of acid per 100 parts of water). All amylaceous substances, converted into dextrin, remain dissolved in the liquid, while the wax, in cooling, forms a crust on the surface. It is taken off and weighed; the difference between its weight and that of the product analyzed will give the quantity of the amylaceous substances.

Flowers of sulphur are recognized readily from the odor of sulphurous acid during combustion on red-hot iron.

Tallow may be detected by the taste and odor. Pure wax has an aromatic, agreeable taste, while that mixed with tallow is repulsive both in taste and smell. Pure wax, worked between the fingers, grows soft, preserving a certain cohesion in all parts. It divides into lumps, which adhere to the fingers, if it is mixed with tallow. The adulteration may also be detected by the thick and nauseating fumes produced when it is burned on heated iron.

Stearic acid may be recognized by means of boiling alcohol, which dissolves it in nearly all proportions and causes it to deposit crystals on cooling, while it is without action on the wax. Blue litmus paper, immersed in alcohol solution, reddens on drying in air, and thus serves for detecting the presence of stearic acid.

Ocher is found by treating the wax with boiling water. A lemon-yellow deposit results, which, taken up with chlorhydric acid, yields with ammonia a lemon-yellow precipitate of ferric oxide.

The powder of burnt bones separates and forms a residue, when the wax is heated with oil of turpentine.

Artificial Beeswax.—This is obtained by mixing the following substances, in approximately the proportions stated: Paraffine, 45 parts, by weight; white Japan vegetable wax, 30 parts, by weight; rosins,

or colophonies, 10 parts, by weight; white pitch, 10 parts, by weight; tallow, 5 parts, by weight; ceresine, colorant, 0.030 parts, by weight; wax perfume, 0.100 parts, by weight. If desired, the paraffine may be replaced with ozokerite, or by a mixture of vaseline and ozokerite, for the purpose of varying the fusing temperature, or rendering it more advantageous for the various applications designed. The following is the method of preparation: Melt on the boiling water bath, shaking constantly, the paraffine, the Japan wax, the rosins, the pitch, and the tallow. When the fusion is complete, add the colorant and the perfume. When these products are perfectly mingled, remove from the fire, allow the mixture to cool, and run it into suitable molds. The wax thus obtained may be employed specially for encaustics for furniture and floors, or for purposes where varnish is employed.

Waxes for Floors, Furniture, etc.—

- I.—White beeswax..... 16 parts
Colophony..... 4 parts
Venice turpentine.. 1 part

Melt the articles together over a gentle fire, and when completely melted and homogeneous, pour into a sizable earthenware vessel, and stir in, while still warm, 6 parts of the best French turpentine. Cool for 24 hours, by which time the mass has acquired the consistence of soft butter, and is ready for use. Its method of use is very simple. It is smeared, in small quantities, on woolen cloths, and with these is rubbed into the wood.

This is the best preparation, but one in which the beeswax is merely dissolved in the turpentine in such a way as to have the consistence of a not too thin oil color, will answer. The wood is treated with this, taking care that the surface is evenly covered with the mixture, and that it does not sink too deeply in the ornaments, corners, etc., of the woodwork. This is best achieved by taking care to scrape off from the cloths all excess of the wax.

If, in the course of 24 hours, the surface is hard, then with a stiff brush go over it, much after the way of polishing a boot. For the corners and angles smaller brushes are used; when necessary, stiff pencils may be employed. Finally, the whole is polished with plush, or velvet rags, in order not to injure the original polish. Give the article a good coat of linseed oil or a washing with petroleum before beginning work.

II.—Articles that are always exposed to the water, floors, doors, especially of oak, should, from time to time, be satu-

rated with oil or wax. A house door, plentifully decorated with wood carving, will not shrink or warp, even where the sun shines hottest on it, when it is frequently treated to saturation with wax and oil. Here a plain dosage with linseed oil is sufficient. Varnish, without the addition of turpentine, should never be used, or if used it should be followed by a coat of wax.

III.—A good floor wax is composed of 2 parts of wax and 3 parts of Venice turpentine, melted on the water bath, and the mixture applied while still hot, using a pencil, or brush, for the application, and when it has become solid and dry, diligently rubbed, or polished down with a woollen cloth, or with a floor brush, especially made for the purpose.

IV.—An emulsion of 5 parts of yellow wax, 2 parts of crude potassium carbonate, and 12 parts of water, boiled together until they assume a milky color and the solids are dissolved, used cold, makes an excellent composition for floors. Any desired color may be given this dressing by stirring in the powdered coloring matter. Use it exactly as described for the first mass.

Gilders' Wax.—For the production of various colorings of gold in fire gilding, the respective places are frequently covered with so-called gilders' wax. These consist of mixtures of various chemicals which have an etching action in the red heat upon the bronze mass, thus causing roughness of unequal depth, as well as through the fact that the composition of the bronze is changed somewhat on the surface, a relief of the gold color being effected in consequence of these two circumstances. The gilding wax is prepared by melting together the finely powdered chemicals with wax according to the following recipes:

	I	II	III	IV	V
Yellow wax.....	32	32	32	96	36
Red chalk.....	3	24	18	48	18
Verdigris.....	2	4	18	32	18
Burnt alum.....	2	4			
Burnt borax.....	—	—	2	1	3
Copper ash.....	—	4	6	20	8
Zinc vitriol.....	—	—	—	32	18
Green vitriol.....	—	—	—	1	6

Grafting Wax.—

1.—Beeswax.....	7	parts
Purified rosin.....	12	parts
Turpentine.....	3	parts
Rape oil.....	1	part
Venice turpentine....	2.5	parts
Zinc white.....	2.5	parts
Color yellow with turmeric.		

II.—Japan wax.....	1	part
Yellow wax.....	3	parts
Rosin.....	8	parts
Turpentine.....	4	parts
Hard paraffine.....	1	part
Suet.....	3	parts
Venice turpentine....	6	parts

Harness Wax.—

Oil of turpentine....	90	parts
Wax, yellow.....	9	parts
Prussian blue.....	1	part
Indigo.....	0.5	parts
Bone black.....	5	parts

Dissolve the wax in the oil by aid of a low heat, on a water bath. Mix the remaining ingredients, which must be well powdered, and work up with a portion of the solution of wax. Finally, add the mixture to the solution, and mix thoroughly on the bath. When a homogeneous liquid is obtained, pour into earthen boxes.

Modeling Wax.—I.—Yellow wax, 1,000 parts; Venice turpentine, 130 parts; lard, 65 parts; bole, 725 parts. The mixture when still liquid is poured into tepid water and kneaded until a plastic mass is obtained.

II.—Summer Modeling Wax.—White wax, 20 parts; ordinary turpentine, 4 parts; sesame oil, 1 part; vermilion, 2 parts.

III.—Winter Modeling Wax.—White wax, 20 parts; ordinary turpentine, 6 parts; sesame oil, 2 parts; vermilion, 2 parts. Preparation same as for Formula I.

Sealing Waxes.—The following formulas may be followed for making sealing wax: Take 4 pounds of shellac, 1 pound of Venice turpentine, and 3 pounds of vermilion. Melt the lac in a copper pan suspended over a clear charcoal fire, then add the turpentine slowly to it, and soon afterwards add the vermilion, stirring briskly all the time with a rod in either hand. In forming the round sticks of sealing wax, a certain portion of the mass should be weighed while it is ductile, divided into the desired number of pieces, and then rolled out upon a warm marble slab by means of a smooth wooden block like that used by apothecaries for rolling a mass of pills.

The oval and square sticks of sealing wax are cast in molds, with the above compound, in a state of fusion. The marks of the lines of junction of the mold box may be afterwards removed by holding the sticks over a clear fire, or passing them over a blue gas flame. Marbled sealing wax is made by mixing

two, three, or more colored kinds together while they are in a semi-fluid state. From the viscosity of the several portions their incorporation is left incomplete, so as to produce the appearance of marbling. Gold sealing wax is made simply by adding gold chrome instead of vermilion into the melted rosins. Wax may be scented by introducing a little essential oil, essence of musk, or other perfume. If 1 part of balsam of Peru be melted along with 99 parts of the sealing-wax composition, an agreeable fragrance will be exhaled in the act of sealing with it. Either lampblack or ivory black serves for the coloring matter of black wax. Sealing wax is often adulterated with rosin, in which case it runs into thin drops at the flame of a candle.

The following mistakes are sometimes made in the manufacture of sealing wax:

I.—Use of filling agents which are too coarsely ground.

II.—Excessive use of filling agents.

III.—Insufficient binding of the pigments and fillings with a suitable adhesive agent, which causes these bodies to absorb the adhesive power of the gums.

IV.—Excessive heating of the mass, caused by improper melting or faulty admixture of the gummy bodies. Turpentine and rosin must be heated before entering the shellac. If this rule is inverted, as is often the case, the shellac sticks to the bottom and burns partly.

Great care must be taken to mix the coloring matter to a paste with spirit or oil of turpentine before adding to the other ingredients. Unless this is done the wax will not be of a regular tint.

Dark Blue Wax.—Three ounces Venetian turpentine, 4 ounces shellac, 1 ounce rosin, 1 ounce Prussian blue, $\frac{1}{2}$ ounce magnesia.

Green Wax.—Two ounces Venetian turpentine, 4 ounces shellac, $1\frac{1}{4}$ ounces rosin, $\frac{1}{2}$ ounce chrome yellow, $\frac{1}{4}$ ounce Prussian blue, 1 ounce magnesia.

Carmine Red Wax.—One ounce Venetian turpentine, 4 ounces shellac, 1 ounce rosin, colophony, $1\frac{1}{4}$ ounces Chinese red, 1 drachm magnesia, with oil of turpentine.

Gold Wax.—Four ounces Venetian turpentine, 8 ounces shellac, 14 sheets of genuine leaf gold, $\frac{1}{2}$ ounce bronze, $\frac{1}{2}$ ounce magnesia, with oil of turpentine.

White Wax.—I.—The wax is bleached by exposing to moist air and to the sun,

but it must first be prepared in thin sheets or ribbons or in grains. For this purpose it is first washed, to free it from the honey which may adhere, melted, and poured into a tin vessel, whose bottom is perforated with narrow slits. The melted wax falls in a thin stream on a wooden cylinder arranged below and half immersed in cold water. This cylinder is turned, and the wax, rolling round in thin leaves, afterwards falls into the water. To melt it in grains, a vessel is made use of, perforated with small openings, which can be rotated. The wax is projected in grains into the cold water. It is spread on frames of muslin, moistened with water several times a day, and exposed to the sun until the wax assumes a fine white. This whiteness, however, is not perfect. The operation of melting and separating into ribbons or grains must be renewed. Finally, it is melted and flowed into molds. The duration of the bleaching may be abridged by adding to the wax, treated as above, from 1.25 to 1.75 per cent of rectified oil of turpentine, free from rosin. In 6 or 8 days a result will be secured which would otherwise require 5 or 6 weeks.

II.—Bleached shellac. . . . 28 parts
Venetian turpentine. . 13 parts
Plaster of Paris. . . . 30 parts

WAX FIGURES RENOVATING:

Apply with a soft stubby brush powdered pumice stone until the wax has lost its yellow hue then with a brush apply a flesh tint consisting of pulverized prepared chalk, powdered pumice stone, and a trifle of oxide of zinc. After this to the same powder just used, mix a little carmine and put on cheeks and lobes of ears. Then with a fine brush, color the lips and inside of angles of the eyes with a liquid solution of carmine to which has been added a trifle of gum arabic.

Weather Forecasters

(See also Hygrometers and Hygroscope.)

I.—It is known that a leaf of blotting paper or a strip of fabric made to change color according to the hygrometric state of the atmosphere has been employed for weather indications in place of a barometer. The following compound is recommended for this purpose: One part of cobalt chloride, 75 parts of nickel oxide, 20 parts of gelatin, and 200 parts of water. A strip of calico, soaked in this solution, will appear green in fine weather, but when moisture intervenes the color disappears.

II.—Copper chloride.... 1 part
Gelatin..... 10 parts
Water..... 100 parts

III.—This is a method of making old-fashioned weather glasses containing a liquid that clouds or solidifies under certain atmospheric conditions:

Camphor..... 2½ drachms
Alcohol..... 11 drachms
Water..... 9 drachms
Saltpeter..... 38 grains
Sal ammoniac.... 38 grains

Dissolve the camphor in the alcohol and the salts in the water and mix the solutions together. Pour in test tubes, cover with wax after corking and make a hole through the cork with a red-hot needle, or draw out the tube until only a pin hole remains. When the camphor, etc., appear soft and powdery, and almost filling the tube, rain with south or southwest winds may be expected; when crystalline, north, northeast, or northwest winds, with fine weather, may be expected; when a portion crystallizes on one side of the tube, wind may be expected from that direction. Fine weather: The substance remains entirely at bottom of tube and the liquid perfectly clear. Coming rain: Substance will rise gradually, liquid will be very clear, with a small star in motion. A coming storm or very high wind: Substance partly at top of tube, and of a leaflike form, liquid very heavy and in a fermenting state. These effects are noticeable 24 hours before the change sets in. In winter: Generally the substance lies higher in the tube. Snow or white frost: Substance very white and small stars in motion. Summer weather: The substance will lie quite low. The substance will lie closer to the tube on the opposite side to the quarter from which the storm is coming. The instrument is nothing more than a scientific toy.

WEATHERPROOFING:

See Paints.

WEED KILLERS:

See Disinfectants.

Weights and Measures

INTERNATIONAL ATOMIC WEIGHTS.

The International Committee on Atomic Weights have presented this table as corrected:

	O=16	H=1
Aluminum..... Al	27.1	26.9
Antimony..... Sb	120.2	119.3
Argon..... A	39.9	39.6

	O=16	H=1
Arsenic..... As	75	74.4
Barium..... Ba	137.4	136.4
Bismuth..... Bi	208.5	206.9
Boron..... B	11	10.9
Bromine..... Br	79.96	79.36
Cadmium..... Cd	112.4	111.6
Cæsium..... Cs	132.9	131.9
Calcium..... Ca	40.1	39.7
Carbon..... C	12	11.01
Cerium..... Ce	140.25	139.2
Chlorine..... Cl	35.45	35.18
Chromium..... Cr	52.1	51.7
Cobalt..... Co	59	58.55
Columbium.... Cb	94	93.3
Copper..... Cu	63.6	63.1
Erbium..... Er	166	164.8
Fluorine..... F	19	18.9
Gadolinium.... Gd	156	154.8
Gallium..... Ga	70	69.5
Germanium.... Ge	72.5	72
Glucinum..... Gl	9.1	9.03
Gold..... Au	197.2	195.7
Helium..... He	4	4
Hydrogen..... H	1.008	1
Indium..... In	115	114.1
Iodine..... I	126.97	126.01
Iridium..... Ir	193	191.5
Iron..... Fe	55.9	55.5
Krypton..... Kr	81.8	81.2
Lanthanum.... La	138.9	137.9
Lead..... Pb	206.9	205.35
Lithium..... Li	7.03	6.98
Magnesium.... Mg	24.36	24.18
Manganese.... Mn	55	54.6
Mercury..... Hg	200	198.5
Molybdenum.. Mo	96	95.3
Neodymium.... Nd	143.6	142.5
Neon..... Ne	20	19.9
Nickel..... Ni	58.7	58.3
Nitrogen..... N	14.04	13.93
Osmium..... Os	191	189.6
Oxygen..... O	16	15.88
Palladium.... Pd	106.5	105.7
Phosphorus... P	31	30.77
Platinum..... Pt	194.8	193.3
Potassium.... K	39.15	38.85
Praseodymium. Pr	140.5	139.4
Radium..... Ra	225	223.3
Rhodium..... Rh	103	102.2
Rubidium.... Rb	85.5	84.9
Ruthenium.... Ru	101.7	100.9
Samarium.... Sm	150.3	149.2
Scandium.... Sc	44.1	43.8
Selenium..... Se	79.2	78.6
Silicon..... Si	28.4	28.2
Silver..... Ag	107.93	107.11
Sodium..... Na	23.05	22.88
Sodium..... Na	87.6	86.94
Strontium.... Sr	87.6	86.94
Sulphur..... S	32.06	31.82
Tantalum.... Ta	183	181.6
Tellurium.... Te	127.6	126.6
Terbium..... Tb	160	158.8
Thallium..... Tl	204.1	202.6

WEIGHTS AND MEASURES

INTERNATIONAL ATOMIC WEIGHTS—Continued.

		O=16	H=1			O=16	H=1
Thorium.....	Th	232.5	230.8	Vanadium.....	V	51.2	50.8
Thulium.....	Tm	171	169.7	Xenon.....	Xe	128	127
Tin.....	Sn	119	118.1	Ytterbium.....	Yb	173	171.7
Titanium.....	Ti	48.1	47.7	Yttrium.....	Yt	89	88.3
Tungsten.....	W	184	182.6	Zinc.....	Zn	65.4	64.9
Uranium.....	U	238.5	236.7	Zirconium.....	Zr	90.6	89.9

UNITED STATES WEIGHTS AND MEASURES

(According to existing standards)

LINEAL

	Inches.	Feet.	Yards.	Rods.	Fur's.	Mile.
12 inches=1 foot.	12=	1				
3 feet=1 yard.	36=	3=	1			
5.5 yards=1 rod.	198=	16.5=	5.5=	1		
40 rods=1 furlong.	7,920=	660=	220=	40=	1	
8 furlongs=1 mile.	63,360=	5,280=	1,760=	320=	8=	1

SURFACE—LAND

	Feet.	Yards.	Rods.	Roods.	Acres.
144 sq. inches=1 square foot.	9	1			
9 square feet=1 square yard.	272.25=	30.25=	1		
30.25 square yards=1 square rod.	10,890=	1,210=	40=	1	
40 square rods=1 square rood.	43,560=	4,840=	160=	4=	1
4 square roods=1 acre.	27,878,400=	3,097,600=	102,400=	2,560=	640
640 acres=1 square mile.					

VOLUME—LIQUID

	Gills.	Pints.	Gallon.	Cub. In.
4 gills=1 pint.	32=	8=	1=	231
2 pints=1 quart.				
4 quarts=1 gallon.				

FLUID MEASURE

Gallon.	Pints.	Ounces.	Drachms.	Minims.	Cubic Centimeters.
1	= 8	= 128	= 1,024	= 61,440	= 3,785.435
	1	= 16	= 128	= 7,680	= 473.179
		1	= 8	= 480	= 29.574
			1	= 60	= 3.697

16 ounces, or a pint, is sometimes called a fluidpound.

TROY WEIGHT

Pound.	Ounces.	Pennyweights.	Grains.	Grams.
1	= 12	= 240	= 5,760	= 373.24
	1	= 20	= 480	= 31.10
		1	= 24	= 1.56

APOTHECARIES' WEIGHT

lb.	3	3	9	gr.	Grams.
Pound.	Ounces.	Drachms.	Scruples.	Grains.	
1	= 12	= 96	= 288	= 5,760	= 373.24
	1	= 8	= 24	= 480	= 31.10
		1	= 3	= 60	= 3.89
			1	= 20	= 1.30
				1	= .06

The pound, ounce, and grain are the same as in Troy weight.

AVOIRDUPOIS WEIGHT

Pound.	Ounces.	Drachms.	Grains (Troy)	Grams.
1	= 16	= 256	= 7,000	= 453.60
	1	= 16	= 437.5	= 28.35
		1	= 27.34	= 1.77

ENGLISH WEIGHTS AND MEASURES

APOTHECARIES' WEIGHT

20 grains	= 1 scruple	= 20 grains
3 scruples	= 1 drachm	= 60 grains
8 drachms	= 1 ounce	= 480 grains
12 ounces	= 1 pound	= 5,760 grains

FLUID MEASURE

60 minims	= 1 fluidrachm
8 drachms	= 1 fluidounce
20 ounces	= 1 pint
8 pints	= 1 gallon

The above weights are usually adopted in formulas.

All chemicals are usually sold by

AVOIRDUPOIS WEIGHT

27 $\frac{1}{2}$ grains	= 1 drachm	= 27 $\frac{1}{2}$ grains
16 drachms	= 1 ounce	= 437 $\frac{1}{2}$ grains
16 ounces	= 1 pound	= 7,000 grains

Precious metals are usually sold by

TROY WEIGHT

24 grains	= 1 pennyweight	= 24 grains
20 pennyweights	= 1 ounce	= 480 grains
12 ounces	= 1 pound	= 5,760 grains

NOTE.—An ounce of metallic silver contains 480 grains, but an ounce of nitrate of silver contains only 437 $\frac{1}{2}$ grains.

WEIGHTS AND MEASURES

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METRIC SYSTEM OF WEIGHTS AND MEASURES

MEASURES OF LENGTH

DENOMINATIONS AND VALUES.		EQUIVALENTS IN USE.
Myriameter.....	10,000 meters	6.2137 miles
Kilometer.....	1,000 meters	.62137 miles, or 3,280 feet, 10 inches
Hectometer.....	100 meters	328 feet and 1 inch
Dekameter.....	10 meters	39.37 inches
Meter.....	1 meter	39.37 inches
Decimeter.....	1-10th of a meter	3.937 inches
Centimeter.....	1-100th of a meter	.3937 inches
Millimeter.....	1-1,000th of a meter	.0394 inches

MEASURES OF SURFACE

DENOMINATIONS AND VALUES.		EQUIVALENTS IN USE.
Hectare.....	10,000 square meters	2.471 acres
Are.....	100 square meters	119.6 square yards
Centare.....	1 square meter	1,550 square inches

MEASURES OF VOLUME

DENOMINATIONS AND VALUES.			EQUIVALENTS IN USE.	
NAMES.	NO. OF LITERS.	CUBIC MEASURES.	DRY MEASURE.	WINE MEASURE.
Kiloliter or stere.....	1,000	1 cubic meter	1.308 cubic yards	264.17 gallons
Hectoliter.....	100	1-10th cubic meter	2 bushels and 3.35 pecks	26.417 gallons
Dekaliter.....	10	10 cubic decimeters	9.08 quarts	2.6417 gallons
Liter.....	1	1 cubic decimeter	.908 quarts	1.0567 quarts
Deciliter.....	1-10	1-10th cubic decimeter	6.1023 cubic inches	.845 gills
Centiliter.....	1-100	10 cubic centimeters	.6102 cubic inches	.338 fluidounces
Milliliter.....	1-1,000	1 cubic centimeter	.061 cubic inches	.27 fluidrachms

WEIGHTS

DENOMINATIONS AND VALUES.			EQUIVALENTS IN USE.
NAMES.	NUMBER OF GRAMS.	WEIGHT OF VOLUME OF WATER AT ITS MAXIMUM DENSITY.	AVOIRDUPOIS WEIGHT.
Millier or Tonneau.....	1,000,000	1 cubic meter	2,204.6 pounds
Quintal.....	100,000	1 hectoliter	220.46 pounds
Myriagram.....	10,000	10 liters	22.046 pounds
Kilogram or Kilo.....	1,000	1 liter	3.5274 ounces
Hectogram.....	100	1 deciliter	.3527 ounces
Dekagram.....	10	10 cubic centimeters	15.432 grains
Gram.....	1	1 cubic centimeter	1.5432 grains
Decigram.....	1-10	1-10th of a cubic centimeter	.1543 grains
Centigram.....	1-100	10 cubic millimeters	.0154 grains
Milligram.....	1-1,000	1 cubic millimeter	

For measuring surfaces, the square dekameter is used under the term of ARE; the hectare, or 100 ares, is equal to about 2½ acres. The unit of capacity is the cubic decimeter or LITER, and the series of measures is formed in the same way as in the case of the table of lengths. The cubic meter is the unit of measure for solid bodies, and is termed STERE. The unit of weight is the GRAM, which is the weight of one cubic centimeter of pure water weighed in a vacuum at the temperature of 4° C. or 39.2° F., which is about its temperature of maximum density. In practice, the term cubic centimeter, abbreviated c.c., is generally used instead of milliliter, and cubic meter instead of kiloliter.

WEIGHTS AND MEASURES

THE CONVERSION OF METRIC INTO ENGLISH WEIGHT

The following table, which contains no error greater than one-tenth of a grain, will suffice for most practical purposes:

1 gram =	15½ grains
2 grams =	30½ grains
3 grams =	46½ grains
4 grams =	61½ grains, or 1 drachm, 1½ grains
5 grams =	77½ grains, or 1 drachm, 17½ grains
6 grams =	92½ grains, or 1 drachm, 32½ grains
7 grams =	108 grains, or 1 drachm, 48 grains
8 grams =	123½ grains, or 2 drachms, 3½ grains
9 grams =	138½ grains, or 2 drachms, 18½ grains
10 grams =	154½ grains, or 2 drachms, 34½ grains
11 grams =	169½ grains, or 2 drachms, 49½ grains
12 grams =	185½ grains, or 3 drachms, 5½ grains
13 grams =	200½ grains, or 3 drachms, 20½ grains
14 grams =	216 grains, or 3 drachms, 36 grains
15 grams =	231½ grains, or 3 drachms, 51½ grains
16 grams =	247 grains, or 4 drachms, 7 grains
17 grams =	262½ grains, or 4 drachms, 22½ grains
18 grams =	277½ grains, or 4 drachms, 37½ grains
19 grams =	293½ grains, or 4 drachms, 53½ grains
20 grams =	308½ grains, or 5 drachms, 8½ grains
30 grams =	463 grains, or 7 drachms, 43 grains
40 grams =	617½ grains, or 10 drachms, 17½ grains
50 grams =	771½ grains, or 12 drachms, 51½ grains
60 grams =	926 grains, or 15 drachms, 26 grains
70 grams =	1,080½ grains, or 18 drachms, 0½ grains
80 grams =	1,234½ grains, or 20 drachms, 34½ grains
90 grams =	1,389 grains, or 23 drachms, 9 grains
100 grams =	1,543½ grains, or 25 drachms, 43½ grains
1,000 grams =	1 kilogram = 32 ounces, 1 drachm, 12½ grains

THE CONVERSION OF METRIC INTO ENGLISH MEASURE

1 cubic centimeter =	17 minims
2 cubic centimeters =	34 minims
3 cubic centimeters =	51 minims
4 cubic centimeters =	68 minims, or 1 drachm, 8 minims
5 cubic centimeters =	85 minims, or 1 drachm, 25 minims
6 cubic centimeters =	101 minims, or 1 drachm, 41 minims
7 cubic centimeters =	118 minims, or 1 drachm, 58 minims
8 cubic centimeters =	135 minims, or 2 drachms, 15 minims
9 cubic centimeters =	152 minims, or 2 drachms, 32 minims
10 cubic centimeters =	169 minims, or 2 drachms, 49 minims
20 cubic centimeters =	338 minims, or 5 drachms, 38 minims
30 cubic centimeters =	507 minims, or 1 ounce, 0 drachm, 27 minims
40 cubic centimeters =	676 minims, or 1 ounce, 3 drachms, 16 minims
50 cubic centimeters =	845 minims, or 1 ounce, 6 drachms, 5 minims
60 cubic centimeters =	1,014 minims, or 2 ounces, 0 drachms, 54 minims
70 cubic centimeters =	1,183 minims, or 2 ounces, 3 drachms, 43 minims
80 cubic centimeters =	1,352 minims, or 2 ounces, 6 drachms, 32 minims
90 cubic centimeters =	1,521 minims, or 3 ounces, 1 drachm, 21 minims
100 cubic centimeters =	1,690 minims, or 3 ounces, 4 drachms, 10 minims
1,000 cubic centimeters =	1 liter = 34 fluidounces nearly, or 2½ pints.

WELDING POWDERS.

See also Steel.

Powder to Weld Wrought Iron at Pale-red Heat with Wrought Iron.—I.—Borax, 1 part (by weight); sal ammoniac, $\frac{1}{2}$ part; water, $\frac{1}{2}$ part. These ingredients are boiled with constant stirring until the mass is stiff; then it is allowed to harden over the fire. Upon cooling, the mass is rubbed up into a powder and mixed with one-third wrought-iron filings free from rust. When the iron has reached red heat, this powder is sprinkled on the parts to be welded, and after it has liquefied, a few blows are sufficient to unite the pieces.

II.—Borax, 2 parts; wrought-iron filings, free from rust, 2 parts; sal ammoniac, 1 part. These pulverized parts are moistened with copaiba balsam and made into a paste, then slowly dried over a fire and again powdered. The application is the same as for Formula I.

Welding Powder to Weld Steel on Wrought Iron at Pale-red Heat.—Borax, 3 parts; potassium cyanide, 2 parts; Berlin blue, 1-100 part. These substances are powdered well, moistened with water; next they are boiled with constant stirring until stiff; then dry over a fire. Upon cooling, the mass is finely pulverized and mixed with 1 part of wrought-iron filings, free from rust. This powder is sprinkled repeatedly upon the hot pieces, and after it has burned in the welding is taken in hand.

WHEEL GREASE:

See Lubricants.

WHETSTONES.

To make artificial whetstones, take gelatin of good quality, dissolve it in equal weight of water, operating in almost complete darkness, and add $1\frac{1}{2}$ per cent of bichromate of potash, previously dissolved. Next take about 9 times the weight of the gelatin employed of very fine emery or fine powdered gun stone, which is mixed intimately with the gelatinized solution. The paste thus obtained is molded into the desired shape, taking care to exercise an energetic pressure in order to consolidate the mass. Finally dry by exposure to the sun.

WHITING:

To Form Masses of Whiting.—Mix the whiting into a stiff paste with water, and the mass will retain its coherence when dry.

Whitewash

(See also Paint.)

Wash the ceiling by wetting it twice with water, laying on as much as can well be floated on, then rub the old color up with a stumpy brush and wipe off with a large sponge. Stop all cracks with whiting and plaster of Paris. When dry, claricole with size and a little of the whitewash when this is dry. If very much stained, paint those parts with turps, color, and, if necessary, claricole again. To make the whitewash, take a dozen pounds of whiting (in large balls), break them up in a pail, and cover with water to soak. During this time melt over a slow fire 4 pounds common size, and at the same time, with a palette knife or small trowel, rub up fine about a dessertspoonful of blue-black with water to a fine paste; then pour the water off the top of the whiting and with a stick stir in the black; when well mixed, stir in the melted size and strain. When cold, it is fit for use. If the jelly is too stiff for use, beat it up well and add a little cold water. Commence whitewashing over the window and so work from the light. Distemper color of any tint may be made by using any other color instead of the blue-black—as ocher, chrome, Dutch pink, raw sienna for yellows and buff; Venetian red, burnt sienna, Indian red or purple brown for reds; celestial blue, ultramarine, indigo for blues; red and blue for purple, gray or lavender; red lead and chrome for orange; Brunswick green for greens.

Ox blood in lime paint is an excellent binding agent for the lime, as it is chiefly composed of albumin, which, like casein or milk, is capable of transforming the lime into casein paint. But the ox blood must be mixed in the lime paint; to use it separately is useless, if not harmful. Whitewashing rough mortar-plastering to saturation is very practical, as it closes all the pores and small holes.

A formula used by the United States Government in making whitewash for light-houses and other public buildings is as follows:

Unslaked lime.....	2 pecks
Common salt.....	1 peck
Rice flour.....	3 pounds
Spanish whiting.....	$\frac{1}{2}$ pound
Glue (clean and white).....	1 pound
Water, a sufficient quantity.	

Slake the lime in a vessel of about 10 gallons capacity; cover it, strain, and add

the salt previously dissolved in warm water. Boil the rice flour in water; soak the glue in water and dissolve on a water bath, and add both, together with the whitening and 5 gallons of hot water to the mixture, stirring all well together. Cover to protect from dirt, and let it stand for a few days, when it will be ready for use. It is to be applied hot, and for that reason should be used from a kettle over a portable furnace.

To Soften Old Whitewash.—Wet the whitewash thoroughly with a wash made of 1 pound of potash dissolved in 10 quarts of water.

WHITEWASH, TO REMOVE:

See Cleaning Preparations and Methods.

WHITE METAL:

See Alloys.

WINDOW-CLEANING COMPOUND:

See Cleaning Compounds.

WINDOW DISPLAY:

See also Sponges.

An attractive window display for stores can be prepared as follows:

In a wide-mouth jar put some sand, say, about 6 inches in depth. Make a mixture of equal parts of aluminum sulphate, copper sulphate, and iron sulphate, coarsely powdered, and strew it over the surface of the sand. Over this layer gently pour a solution of sodium silicate, dissolved in 3 parts of hot water, taking care not to disturb the layer of sulphates. In about a week or 10 days the surface will be covered with crystals of different colors, being silicates of different metals employed. Now take some pure water and let it run into the vessel by a small tube, using a little more of it than you used of the water-glass solution. This will displace the water-glass solution, and a fresh crop of crystals will come in the silicates, and makes, when properly done, a pretty scene. Take care in pouring in the water to let the point of the tube be so arranged as not to disturb the crop of silicates.

WINDOW PERFUME.

In Paris an apparatus has been introduced consisting of a small tube which is attached lengthwise on the exterior of the shop windows. Through numerous little holes a warm, lightly perfumed current of air is passed, which pleasantly tickles the olfactory nerves of the looker-on and at the same time keeps the panes

clear and clean, so that the goods exhibited present the best possible appearance.

WINE BRICKS:

Simply press dry grapes into a brick with enough grape sugar or glucose added to hold it together. The drying is not carried as far as is the case with raisins or prunes. Raisins or prunes would do as well for the purpose.

Wines and Liquors

BITTERS.

Bitters, as the name indicates, are merely tinctures of bitter roots and barks, with the addition of spices to flavor, and depend for their effect upon their tonic action on the stomach. Taken too frequently, however, they may do harm, by overstimulating the digestive organs.

The recipes for some of these preparations run to great lengths, one for Angostura bitters containing no fewer than 28 ingredients. A very good article, however, may be made without all this elaboration. The following, for instance, make a very good preparation:

Gentian root (sliced) ..	12 ounces
Cinnamon bark	10 ounces
Caraway seeds	10 ounces
Juniper berries	2 ounces
Cloves	1 ounce
Alcohol, 90 per cent ..	7 pints

Macerate for a week; strain, press out, and filter, then add

Capillaire	1½ pints
Water to make up ..	2½ gallons

Strength about 45 u. p.

Still another formula calls for Angostura bark, 2½ ounces; gentian root, 1 ounce; cardamom seeds, ½ ounce; Turkey rhubarb, ½ ounce; orange peel, 4 ounces; caraways, ½ ounce; cinnamon bark, ½ ounce; cloves, ¼ ounce.

Brandy Bitters.—

Sliced gentian root. . .	3 pounds
Dried orange peel. . . .	2 pounds
Cardamom seed	1 pound
Bruised cinnamon	½ pound
Cochineal	2 ounces
Brandy	10 pints

Macerate for 14 days and strain.

Hostetter's Bitters.—

Calamus root	1 pound
Orange peel	1 pound
Peruvian bark	1 pound
Gentian root	1 pound

Calumba root.....	1 pound
Rhubarb root.....	4 ounces
Cinnamon bark.....	2 ounces
Cloves.....	1 ounce
Diluted alcohol.....	2 gallons
Water.....	1 gallon
Sugar.....	1 pound

Macerate together for 2 weeks.

CORDIALS.

Cordials, according to the *Spatula*, are flavored liquors containing from 40 to 50 per cent of alcohol (from 52 to 64 fluidounces to each gallon) and from 20 to 25 per cent of sugar (from 25 to 32 ounces avoirdupois to each gallon).

Cordials, while used in this country to some degree, have their greatest consumption in foreign lands, especially in France and Germany.

Usually such mixtures as these are clarified or "fined" only with considerable difficulty, as the finally divided particles of oil pass easily through the pores of the filter paper. Purified talcum will be found to be an excellent clarifying medium; it should be agitated with the liquid and the liquid then passed through a thoroughly wetted filter. The filtrate should be returned again and again to the filter until it filters perfectly bright. Purified talcum being chemically inert is superior to magnesium carbonate and other substances which are recommended for this purpose.

When the filtering process is completed the liquids should at once be put into suitable bottles which should be filled and tightly corked and sealed. Wrap the bottles in paper and store away, laying the bottles on their sides in a moderately warm place. A shelf near the ceiling is a good place. Warmth and age improve the beverages, as it appears to more perfectly blend the flavors, so that the older the liquor becomes the better it is. These liquids must never be kept in a cold place, as the cold might cause the volatile oils to separate.

The following formulas are for the production of cordials of the best quality, and therefore only the very best of materials should be used; the essential oils should be of unquestionable quality and strictly fresh, while the alcohol must be free from fusel oil, the water distilled, and the sugar white, free from bluing, and if liquors of any kind should be called for in any formula only the very best should be used. The oils and other flavoring substances should be dissolved in the alcohol and the sugar in the water. Then mix the two solutions and filter clear.

Alkermes Cordial.—

Mace.....	1½ avoirdupois ounces
Ceylon cinnamon.....	1½ avoirdupois ounces
Cloves.....	¼ avoirdupois ounce
Rose water	
(best).....	6 fluidounces
Sugar.....	28 avoirdupois ounces
Deodorized alcohol.....	52 fluidounces
Distilled water,	
q. s.....	1 gallon

Reduce the mace, cinnamon, and cloves to a coarse powder, macerate with the alcohol for several days, agitating occasionally, then add the remaining ingredients, and filter clear.

Anise Cordial.—

Anethol.....	7 fluidrachms
Oil of fennel seed..	80 minims
Oil of bitter	
almonds.....	16 drops
Deodorized alcohol	8 pints
Simple syrup.....	5 pints
Distilled water, q. s.	16 pints

Mix the oils and anethol with the alcohol and the syrup with the water; mix the two and filter clear, as directed.

Blackberry Cordial.—This beverage is usually misnamed "blackberry brandy" or "blackberry wine." This latter belongs only to wines obtained by the fermentation of the blackberry juice. When this is distilled then a true blackberry brandy is obtained, just as ordinary brandy is obtained by distilling ordinary wines.

The name is frequently applied to a preparation containing blackberry root often combined with other astringents, but the true blackberry cordial is made according to the formulas given herewith. Most of these mention brandy, and this article should be good and fusel free, or it may be replaced by good whisky, or even by diluted alcohol, depending on whether a high-priced or cheap cordial is desired.

I.—Fresh blackberry juice, 3 pints; sugar, 7½ ounces; water, 30 fluidounces; brandy, 7½ pints; oil of cloves, 3 drops; oil of cinnamon, 3 drops; alcohol, 6 fluidrachms. Dissolve the sugar in the water and juice, then add the liquor. Dissolve the oils in the alcohol and add ½ to the first solution, and if not sufficiently flavored add more of the second solution. Then filter.

II.—Fresh blackberry juice, 4 pints; powdered nutmeg (fresh), 1 ounce; powdered cinnamon (fresh), 1 ounce; powdered pimento (fresh), ½ ounce; powdered cloves

(fresh), $\frac{1}{2}$ ounce; brandy, $2\frac{1}{2}$ pints; sugar, $2\frac{1}{2}$ pounds. Macerate the spices in the brandy for several days. Dissolve the sugar in the juice and mix and filter clear.

Cherry Cordials.—

I.—Oil of bitter almonds	8 drops
Oil of cinnamon.....	1 drop
Oil of cloves.....	1 drop
Acetic ether.....	12 drops
Ceanthie ether.....	1 drop
Vanilla extract.....	1 drachm
Alcohol.....	3 pints
Sugar.....	3 pounds
Cherry juice.....	20 ounces
Distilled water, q. s..	1 gallon

The oils, ethers, and extracts must be dissolved in the alcohol, the sugar in part of the water, then mix, add the juice and filter clear. When the juice is not sufficiently sour, add a small amount of solution of citric acid. To color, use caramel.

II.—Vanilla extract.....	10 drops
Oil of cinnamon.....	10 drops
Oil of bitter almonds.	10 drops
Oil of cloves.....	3 drops
Oil of nutmeg.....	3 drops
Alcohol.....	$2\frac{1}{2}$ pints
Cherry juice.....	$2\frac{1}{2}$ pints
Simple syrup.....	3 pints

Dissolve the oils in the alcohol, then add the other ingredients and filter clear. It is better to make this cordial during the cherry season so as to obtain the fresh expressed juice of the cherry.

Curacao Cordials.—

I.—Curacao orange peel..	6 ounces
Cinnamon.....	$\frac{3}{4}$ ounce
Mace.....	$2\frac{1}{2}$ drachms
Alcohol.....	$3\frac{1}{2}$ pints
Water.....	$4\frac{1}{2}$ pints
Sugar.....	12 ounces

Mix the first three ingredients and reduce them to a coarse powder, then mix with the alcohol and 4 pints of water and macerate for 8 days with an occasional agitation, express, add the sugar and enough water to make a gallon of finished product. Filter clear.

II.—Curacao or bitter orange peel.....	2 ounces
Cloves.....	80 grains
Cinnamon.....	80 grains
Cochineal.....	60 grains
Oil of orange (best)..	1 drachm
Orange-flower water.	$\frac{1}{2}$ pint
Holland gin.....	1 pint
Alcohol.....	2 pints
Sugar.....	3 pints
Water, q. s.....	1 gallon

Reduce the solids to a coarse powder, add the alcohol and macerate 3 days. Then add the oil, gin, and 3 pints of water and continue the maceration for 8 days more, agitating once a day, strain and add sugar dissolved in balance of the water. Then add the orange-flower water and filter.

Kola Cordial.—

Kola nuts, roasted and powdered.....	7 ounces
Cochineal powder....	30 grains
Extract of vanilla....	3 drachms
Arrac.....	3 ounces
Sugar.....	7 pounds
Alcohol.....	6 pints
Water, distilled.....	6 pints

Macerate kola and cochineal with alcohol for 10 days, agitate daily, add arrac, vanilla, and sugar dissolved in water. Filter.

Kümmel Cordials.—

I.—Oil of caraway.....	30 drops
Oil of peppermint....	3 drops
Oil of lemon.....	3 drops
Acetic ether.....	30 drops
Spirit of nitrous ether	30 drops
Sugar.....	72 ounces
Alcohol.....	96 ounces
Water.....	96 ounces

Dissolve the oils and ethers in the alcohol, and the sugar in the water. Mix and filter.

II.—Oil of caraway.....	20 drops
Oil of sweet fennel...	2 drops
Oil of cinnamon.....	1 drop
Sugar.....	14 ounces
Alcohol.....	2 pints
Water.....	4 pints

Prepare as in Formula I.

Orange Cordials.—Many of the preparations sold under this name are not really orange cordials, but are varying mixtures of uncertain composition, possibly flavored with orange. The following are made by the use of oranges:

I.—Sugar.....	8 avoirdupois pounds
Water.....	$2\frac{1}{4}$ gallons
Oranges....	15

Dissolve the sugar in the water by the aid of a gentle heat, express the oranges, add the juice and rinds to the syrup, put the mixture into a cask, keep the whole in a warm place for 3 or 4 days, stirring frequently, then close the cask, set aside in a cool cellar and draw off the clear liquid.

II.—Express the juice from sweet oranges, add water equal to the volume

of juice obtained, and macerate the expressed oranges with the juice and water for about 12 hours. For each gallon of juice, add 1 pound of granulated sugar, grape sugar, or glucose, put the whole into a suitable vessel, covering to exclude the dust, place in a warm location until fermentation is completed, draw off the clear liquid, and preserve in well-stoppered stout bottles in a cool place.

III.—Orange wine suitable for "soda" purposes may be prepared by mixing 3 fluidounces of orange essence with 13 fluidounces of sweet Catawba or other mild wine. Some syrup may be added to this if desired.

Rose Cordial.—

Oil of rose, very best.. 3 drops
 Palmarosa oil..... 3 drops
 Sugar..... 28 ounces
 Alcohol..... 52 ounces
 Distilled water, q. s.. 8 pints

Dissolve the sugar in the water and the oils in the alcohol; mix the solutions, color a rose tint, and filter clear.

Spearmint Cordial.—

Oil of spearmint.... 30 drops
 Sugar..... 28 ounces
 Alcohol..... 52 ounces
 Distilled water, q. s.. 8 pints

Dissolve the sugar in the water and the oil in the alcohol; mix the two solutions, color green, and filter clear.

Absinthe.—

I.—Oil of wormwood... 96 drops
 Oil of star anise... 72 drops
 Oil of aniseed.... 48 drops
 Oil of coriander... 48 drops
 Oil of fennel, pure. 48 drops
 Oil of angelica
 root..... 24 drops
 Oil of thyme..... 24 drops
 Alcohol (pure)... 162 fluidounces
 Distilled water... 30 fluidounces

Dissolve the oils in the alcohol, add the water, color green, and filter clear.

II.—Oil of wormwood.. 36 drops
 Oil of orange peel. 30 drops
 Oil of star anise... 12 drops
 Oil of neroli petate. 5 drops
 Fresh oil of lemon. 9 drops
 Acetic ether..... 24 drops
 Sugar..... 30 avoirdupois
 ounces
 Alcohol, deodorized 90 fluidounces
 Distilled water.... 78 fluidounces

Dissolve the oils and ether in the alcohol and the sugar in the water; then mix thoroughly, color green, and filter clear.

DETANNATING WINE.

According to Caspari, the presence of appreciable quantities of tannin in wine is decidedly objectionable if the wine is to be used in connection with iron and other metallic salts; moreover, tannin is incompatible with alkaloids, and hence wine not deprived of its tannin should never be used as a menstruum for alkaloidal drugs. The process of freeing wines from tannin is termed detannation, and is a very simple operation. The easiest plan is to add $\frac{1}{2}$ ounce of gelatin in number 40 or number 60 powder to 1 gallon of the wine, to agitate occasionally during 24 or 48 hours, and then to filter. The operation is preferably carried out during cold weather or in a cold apartment, as heat will cause the gelatin to dissolve, and the maceration must be continued until a small portion of the wine mixed with a few drops of ferric chloride solution shows no darkening of color. Gelatin in large pieces is not suitable, especially with wines containing much tannin, since the newly formed tannate of gelatin will be deposited on the surface and prevent further intimate contact of the gelatin with the wine. Formerly freshly prepared ferric hydroxide was much employed for detannating wine, but the chief objection to its use was due to the fact that some iron invariably was taken up by the acid present in the wine; moreover, the process was more tedious than in the case of gelatin. As the removal of tannin from wine in no way interferes with its quality—alcoholic strength and aroma remaining the same, and only coloring matter being lost—a supply of detannated wine should be kept on hand, for it requires very little more labor to detannate a gallon than a pint.

If ferric hydroxide is to be used, it must be freshly prepared, and a convenient quantity then be added to the wine—about 8 ounces of the expressed, but moist, precipitate to a gallon.

PREVENTION OF FERMENTATION.

Fermentation may be prevented in either of two ways:

(1) By chemical methods, which consist in the addition of germ poisons or antiseptics, which either kill the germs or prevent their growth. Of these the principal ones used are salicylic, sulphurous, boracic, and benzoic acids, formalin, fluorides, and saccharine. As these substances are generally regarded as adulterants and injurious, their use is not recommended.

(2) The germs are either removed by

some mechanical means such as a filtering or a centrifugal apparatus, or they are destroyed by heat or electricity. Heat has so far been found the most practical.

When a liquid is heated to a sufficiently high temperature all organisms in it are killed. The degree of heat required, however, differs not only with the particular kind of organism, but also with the liquid in which it is held. Time is also a factor. An organism may not be killed if heated to a high temperature and quickly cooled. If, however, the temperature is kept at the same high degree for some time, it will be killed. It must also be borne in mind that fungi, including yeasts, exist in the growing and the resting states, the latter being much more resistant than the former. One characteristic of the fungi and their spores is their great resistance to heat when dry. In this state they can be heated to 212° F. without being killed. The spores of the common mold are even more resistant. This should be well considered in sterilizing bottles and corks, which should be steamed to 240° F. for at least 15 minutes.

Practical tests so far made indicate that grape juice can be safely sterilized at from 165° to 176° F. At this temperature the flavor is hardly changed, while at a temperature much above 200° F. it is. This is an important point, as the flavor and quality of the product depend on it.

Use only clean, sound, well-ripened, but not over-ripe grapes. If an ordinary cider mill is at hand, it may be used for crushing and pressing, or the grapes may be crushed and pressed with the hands. If a light-colored juice is desired, put the crushed grapes in a cleanly washed cloth sack and tie up. Then either hang up securely and twist it or let two persons take hold, one on each end of the sack and twist until the greater part of the juice is expressed. Next gradually heat the juice in a double boiler or a large stone jar in a pan of hot water, so that the juice does not come in direct contact with the fire at a temperature of 180° to 200° F., never above 200° F. It is best to use a thermometer, but if there be none at hand heat the juice until it steams, but do not allow it to boil. Put it in a glass or enameled vessel to settle for 24 hours; carefully drain the juice from the sediment, and run it through several thicknesses of clean flannel, or a conic filter made from woolen cloth or felt may be used. This filter is fixed to a hoop of iron, which

can be suspended wherever necessary. After this fill into clean bottles. Do not fill entirely, but leave room for the liquid to expand when again heated. Fit a thin board over the bottom of an ordinary wash boiler, set the filled bottles (ordinary glass fruit jars are just as good) in it, fill in with water around the bottles to within about an inch of the tops, and gradually heat until it is about to simmer. Then take the bottles out and cork or seal immediately. It is a good idea to take the further precaution of sealing the corks over with sealing wax or paraffine to prevent mold germs from entering through the corks. Should it be desired to make red juice, heat the crushed grapes to not above 200° F., strain through a clean cloth or drip bag (no pressure should be used), set away to cool and settle, and proceed the same as with light-colored juice. Many people do not even go to the trouble of letting the juice settle after straining it, but re-heat and seal it up immediately, simply setting the vessel away in a cool place in an upright position where they will be undisturbed. The juice is thus allowed to settle, and when wanted for use the clear juice is simply taken off the sediment. Any person familiar with the process of canning fruit can also preserve grape juice, for the principles involved are identical.

One of the leading defects so far found in unfermented juice is that much of it is not clear, a condition which very much detracts from its otherwise attractive appearance, and due to two causes already alluded to. Either the final sterilization in bottles has been at a higher temperature than the preceding one, or the juice has not been properly filtered or has not been filtered at all. In other cases the juice has been sterilized at such a high temperature that it has a disagreeable scorched taste. It should be remembered that attempts to sterilize at a temperature above 195° F. are dangerous so far as the flavor of the finished product is concerned.

Another serious mistake is sometimes made by putting the juice into bottles so large that much of it becomes spoiled before it is used after the bottles are opened. Unfermented grape juice properly made and bottled will keep indefinitely, if it is not exposed to the atmosphere or mold germs; but when a bottle is once opened it should, like canned goods, be used as soon as possible to keep from spoiling.

Another method of making unfermented grape juice, which is often re-

sorted to where a sufficiently large quantity is made at one time, consists in this:

Take a clean keg or barrel (one that has previously been made sweet). Lay this upon a skid consisting of two scantlings or pieces of timber of perhaps 20 feet long, in such a manner as to make a runway. Then take a sulphur match, made by dipping strips of clean muslin about 1 inch wide and 10 inches long into melted brimstone, cool it and attach it to a piece of wire fastened in the lower end of a bung and bent over at the end, so as to form a hook. Light the match and by means of the wire suspend it in the barrel, bung the barrel up tight, and allow it to burn as long as it will. Repeat this until fresh sulphur matches will no longer burn in the barrel.

Then take enough fresh grape juice to fill the barrel one-third full, bung up tight, roll and agitate violently on the skid for a few minutes. Next burn more sulphur matches in it until no more will burn, fill in more juice until the barrel is about two-thirds full; agitate and roll again. Repeat the burning process as before, after which fill the barrel completely with grape juice and roll. The barrel should then be bunged tightly and stored in a cool place with the bung up, and so secured that the package cannot be shaken. In the course of a few weeks the juice will have become clear and can then be racked off and filled into bottles or jars direct, sterilized, and corked or sealed up ready for use. By this method, however, unless skillfully handled, the juice is apt to have a slight taste of the sulphur.

The following are the component parts of a California and a Concord unfermented grape juice:

	Concord Per Cent	California Per Cent
Solid contents.	20.37	20.60
Total acids (as tartaric).663	.53
Volatile acids.023	.03
Grape sugar.	18.54	19.15
Free tartaric acids.025	.07
Ash.255	.19
Phosphoric acids.027	.04
Cream of tartar.55	.59

This table is interesting in so far that the California unfermented grape juice was made from *Vinifera* or foreign varieties, whereas the Concord was a *Labruska* or one of the American sorts. The difference in taste and smell is even more pronounced than the analysis would indicate.

Small quantities of grape juice may be preserved in bottles. Fruit is likely to be dusty and to be soiled in other ways, and grapes, like other fruits, should be well washed before using. Leaves or other extraneous matter should also be removed. The juice is obtained by moderate pressure in an ordinary screw press, and strained through felt. By gently heating, the albuminous matter is coagulated and may be skimmed off, and further clarification may be effected by filtering through paper, but such filtration must be done as rapidly as possible, using a number of filters and excluding the air as much as possible.

The juice so obtained may be preserved by sterilization, in the following manner: Put the juice in the bottles in which it is to be kept, filling them very nearly full; place the bottles, unstoppered, in a kettle filled with cold water, so arranging them on a wooden perforated "false bottom" or other like contrivance as to prevent their immediate contact with the metal, this preventing unequal heating and possible fracture. Now heat the water, gradually raising the temperature to the boiling point, and maintain at that until the juice attains a boiling temperature; then close the bottles with perfectly fitting corks, which have been kept immersed in boiling water for a short time before use.

The corks should not be fastened in any way, for, if the sterilization is not complete, fermentation and consequent explosion of the bottle may occur unless the cork should be forced out.

If the juice is to be used for syrup, as for use at the soda fountain, the best method is to make a concentrated syrup at once, using about 2 pounds of refined sugar to 1 pint of juice, dissolving by a gentle heat. This syrup may be made by simple agitation without heat; and a finer flavor thus results, but its keeping quality would be uncertain.

The juices found in the market are frequently preserved by means of antiseptics, but so far none have been proposed for this purpose which can be considered entirely wholesome. Physiological experiments have shown that while bodies suited for this purpose may be apparently without bad effect at first, their repeated ingestion is likely to cause gastric disturbance.

SPARKLING WINES.

An apparatus for converting still into foaming wines, and doing this efficiently, simply, and rapidly, consists of a vertical steel tube, which turns on an axis, and

bears several adjustable glass globes that are in connection with each other by means of distributing valves, the latter being of silver-plated bronze. The glass globes serve as containers for carbonic acid, and are kept supplied with this gas from a cylinder connected therewith.

The wine to be impregnated with the acid is taken from a cask, through a special tube, which also produces a light pressure of carbonic acid on the cask, the object of which is to prevent the access of atmospheric air to the wine within, and, besides, to cause the liquid to pass into the bottle without jar or stroke. The bottles stand under the distributing valves, or levers, placed above and below them. Now, if the cock, by means of which the glass bulbs and the bottles are brought into connection, is slightly opened, and the desired lever is put in action, the carbonic acid at once forces the air out of the bottles, and sterilizes them. The upper bottles are now gradually filled. The whole apparatus, including the filled bottles, is now tilted over, and the wine, of its own weight, flows through collectors filled with carbonic acid, and passes, impregnated with the gas, into other bottles placed below. Each bottle is filled in course, the time required for each being some 45 seconds. The saturation of the liquid with carbonic acid is so complete and plentiful that there is no need of hurry in corking.

By means of this apparatus any desired still wine is at once converted into a sparkling one, preserving at the same time its own peculiarities of taste, bouquet, etc. The apparatus may be used equally well upon fruit juices, milk, and, in fact, any kind of liquid, its extreme simplicity permitting of easy and rapid cleansing.

ARTIFICIAL FRENCH BRANDY.

I.—The following is Eugene Dieterich's formula for *Spiritus vini Gallici artificialis*:

Tincture of gall-apples.....	10 parts
Aromatic tincture...	5 parts
Purified wood vinegar.....	5 parts
Spirit of nitrous ether	10 parts
Acetic ether.....	1 part
Alcohol, 68 per cent.	570 parts
Distilled water.....	400 parts

Mix, adding the water last, let stand for several days, then filter.

II.—The *Münchener Apotheker Verein* has adopted the following formula for the same thing:

Acetic acid, dilute, 90 per cent.....	4 parts
Acetic ether.....	4 parts
Tincture aromatic..	40 parts
Cognac essence.....	40 parts
Spirit of nitrous ether.....	20 parts
Alcohol, 90 per cent.	5,000 parts
Water, distilled.....	2,500 parts

Add the acids, ethers, etc., to the alcohol, and finally add the water. Let stand several days, and, if necessary, filter.

III.—The Berlin Apothecaries have adopted the following as a magistral formula:

Aromatic tincture...	4 parts
Spirit of nitrous ether.....	5 parts
Alcohol, 90 per cent.	1,000 parts
Distilled water, quantity sufficient to make.....	2,000 parts

Mix the tincture and ether with the alcohol, add the water and for every ounce add one drop of tincture of rhatany.

Of these formulas the first is to be preferred as a close imitation of the taste of the genuine article. To imitate the color use burnt sugar.

LIQUEURS.

Many are familiar with the properties of liqueurs but believe them to be very complex and even mysterious compounds. This is, of course, due to the fact that the formulas are of foreign origin and many of them have been kept more or less secret for some time. Owing to the peculiar combination of the bouquet oils and flavors, it is impossible to make accurate analyses of them. But by the use of formulas now given, these products seem to be very nearly duplicated.

It is necessary to use the best sugar and oils obtainable in the preparation of the liqueurs. As there are so many grades of essential oils on the market, it is difficult to obtain the best indirectly. The value of the cordials is enhanced by the richness and odor and flavor of the oils, so only the best qualities should be used.

For filtering, flannel or felt is valuable. Flannel is cheaper and more easily washed. It is necessary to return filtrate several times with any of the filtering media.

As a clarifying agent talcum allowed to stand several days acts well. These rules are common to all.

The operations are all simple:
First: Heat all mixtures. Second:
Keep the product in the dark. Third:
Keep in warm place.
The liqueurs are heated to ripen the
bouquet flavor, it having effect similar to
age. To protect the ethereal oils, air and
light are excluded; hence it is recom-
mended that the bottles be filled to the
stopper. The liqueurs taste best at a
temperature not exceeding 55° F. They
are all improved with age, especially
many of the bouquet oils.

Bénédictine.—

I.—Bitter almonds..	40	grams
Powdered nut- meg.....	4.500	grams
Extract vanilla..	120	grams
P o w d e r e d cloves.....	2	grams
Lemons, sliced..	2	grams
True saffron....	.600	grams
Sugar.....	2,000	grams
Boiling milk....	1,000	c.c.
Alcohol, 95 per cent.....	2,000	c.c.
Distilled water..	2,500	c.c.

Mix. Let stand 9 days with occasional
agitation. Filter sufficiently.

II.—Essence Béné- dictine.....	75	c.c.
Alcohol, 95 per cent.....	1,700	c.c.

Mix.

Sugar.....	1,750	grams
Water, distilled.	1,600	c.c.

Mix together, when clear solution of
sugar is obtained. Color with caramel.
Filter sufficiently.

NOTE.—This liqueur should be at least
1 year old before used.

**Essence Bénédictine for Bénédictine
No. II.—**

I.—Myrrh.....	1	part
Decorticated carda- mom.....	1	part
Mace.....	1	part
Ginger.....	10	parts
Galanga root.....	10	parts
Orange peel (cut)..	10	parts
Extract aloe.....	4	parts
Alcohol.....	160	parts
Water.....	80	parts

Mix, macerate 10 days and filter.

II.—Extract licorice....	20	parts
Sweet spirits niter..	200	parts
Acetic ether.....	30	parts
Spirits ammonia....	1	part
Coumarin.....	.12	parts
Vanillin.....	1	part

III.—Oil lemon.....	3	drops
Oil orange peel....	3	drops
Oil wormwood.....	2.5	drops
Oil galanga.....	2	drops
Oil ginger.....	1	drop
Oil anise.....	15	drops
Oil cascarilla.....	15	drops
Oil bitter almond..	12	drops
Oil milfoil.....	10	drops
Oil sassafras.....	7	drops
Oil angelica.....	6	drops
Oil hyssop.....	4	drops
Oil cardamom.....	2	drops
Oil hops.....	2	drops
Oil juniper.....	1	drop
Oil rosemary.....	1	drop

Mix A, B, and C.

NOTE.—This essence should stand 2
years before being used for liqueurs.

**Chartreuse.—I.—Elixir végétal de la
Grande Chartreuse.**

Fresh balm mint herbs.....	64	parts
Fresh hyssop herbs..	64	parts
Angelica herbs and root, fresh, together	32	parts
Cinnamon.....	16	parts
Saffron.....	4	parts
Mace.....	4	parts

Subject the above ingredients to
maceration for a week with alcohol (96
per cent), 1,000 parts, then squeeze off
and distill the liquid obtained over a
certain quantity of fresh herbs of balm
and hyssop. After 125 parts of sugar
have been added to the resultant liqueur,
filter.

The genuine Chartreuse comes in
three different colors, viz., green, white,
and yellow. The coloration, however,
is not artificial, but is determined by the
addition of varying quantities of fresh
herbs in the distillation. But since it
would require long and tedious trials to
produce the right color in a small manu-
facture, the yellow shade is best im-
parted by a little tincture of saffron, and
the green one by the addition of a few
drops of indigo solution.

II.—Eau des Carmes....	3½	ounces
Alcohol.....	1	quart
Distilled water.....	1	quart
Sugar.....	1½	pounds
Tincture of saffron...	1	ounce

Mix. Dissolve sugar in warm water,
cool, strain, add remainder of in-
gredients, and filter. This is known as
yellow Chartreuse.

Curaçao Liqueur.—

A.—Oil lemon, q. s.	10 drops
Oil bitter almond, q. s.	5 drops
Oil curaçoa orange....	15 parts
Oil sweet orange.....	1 part
Oil bitter orange.....	1 part
Cochineal.....	1 part
French brandy.....	50 parts

B.—Alcohol.....4,500 parts

C.—Sugar.....3,500 parts
Water (distilled).....4,000 parts

Mix A, B, and C. Filter. Color with caramel.

May Bowl or May Wine.—The principal ingredient of May bowl, or that which gives it its flavor and bouquet, is fresh *Waldmeisterkraut* (*Asperula odorata*), the "woodruff" or "sweet grass," "star grass," and a dozen other aliases, of a plant growing wild all over Europe, both continental and insular, and cultivated by some gardeners in this country. It is accredited with being a diuretic, deobstruent and hepatic stimulant, of no mean order, though it has long been banished from the pharmacopœia.

In Baden and in Bavaria in preparing *Maitrank* the practice was formerly to first make an essence—*Maitrankessenz*, for the preparation of which every housewife had a formula of her own. The following was that generally used in the south of Germany:

- I.—Fresh, budding
woodruff, cut fine 500 parts
Alcohol, commercial (90 per cent). 1,000 parts

Digest together for 14 days, then filter and press off. Many add to this some flavoring oil. As coumarin has been found to be the principle to which the *Waldmeister* owes its odor, many add to the above Tonka bean, chopped fine, 1 part to the thousand. From about 12 to 15 drachms of this essence is added to make a gallon of the wine, which has about the following formula:

French brandy, say	4 drachms
Oil of unripe oranges.....	80 drops
Sugar.....	4 to 8 ounces
Essence.....	12 drachms
Wine to make.....	1 gallon

II.—Take enough good woodruff (*Waldmeister*) of fine aroma and flavor. Remove all parts that will not add to the excellence of the product, such as wilted, dead, or imperfect leaves, stems, etc., and wash the residue thoroughly in cold water, and with as little pressure as possible. Now choose a flask with a neck

sufficiently wide to receive the stems without pressing or bruising them, and let the pieces fall into it. Pour in sufficient strong alcohol (96 per cent) to cover the herbs completely. In from 30 to 40 minutes the entire aroma is taken up by the alcohol, which takes on a beautiful green color, which, unfortunately, does not last, disappearing in a few days, but without affecting the aroma in the least. The alcohol should now be poured off, for if left to macerate longer, while it would gain in aroma, it will also take up a certain bitter principle that detracts from the delicacy of flavor and aroma. The extract is now poured on a fresh quantity of the herb, and continue proceeding in this manner until a sufficiently concentrated extract is obtained to give aroma to 100 times its weight of wine or cider.

III.—Fresh woodruff, in bloom or flower, is freed from the lower part of its stem and leaves, and also of all foreign or inert matter. The herb is then lightly stuck into a wide-mouth bottle, and covered with strong alcohol. After 30 minutes pour off the liquor on fresh woodruff. In another half hour the essence is ready, though it should not be used immediately. It should be kept at cellar heat (about 60° F.) for a few days, or until the green color vanishes. Any addition to the essence of aromatics, such as orange peel, lemons, spices, etc., is to be avoided. To prepare the *Maitrank*, add the essence to any good white wine, tasting and testing, until the flavor suits.

The following are other formulas for the drink:

- IV.—Good white wine or
cider..... 65 parts
Alcohol, dilute..... 20 parts
Sugar..... 10 parts
Maitrankessenz..... 1 part

Mix.

Maraschino Liqueur.—

Oil bitter almonds....	15 minims
Essence vanilla.....	1 drachm
Jasmine extract.....	2 drops
Raspberry essence....	10 drops
Oil neroli.....	10 drops
Oil lemon.....	15 minims
Spirits nitrous ether..	2 drachms
Alcohol.....	6 pints
Sugar.....	8 pounds
Rose water.....	10 ounces
Water sufficient to make.....	2 gallons

Make a liquor in the usual manner.

To Clarify Liqueurs.—For the clarification of turbid liqueurs, burnt pow-

dered alum is frequently employed. Make a trial with 200 parts of the dim liqueur, to which 1.5 parts of burnt powdered alum is added; shake well and let stand until the liquid is clear. Then decant and filter the last portion. If the trial is successful, the whole stock may be clarified in this manner.

MEDICINAL WINES:

Beef and Iron.—The following formula is recommended by the American Pharmaceutical Association:

I.—Extract of beef.....	35 grams
Tincture of citro-chloride of iron..	35 c.c.
Compound spirit of orange.....	1 c.c.
Hot water.....	60 c.c.
Alcohol.....	125 c.c.
Syrup.....	125 c.c.
Sherry wine sufficient to make....	1,000 c.c.

Rub the extract of beef with the hot water, and add, while stirring, the alcohol. Allow to stand 3 days or more, then filter and distill off the alcohol. Add to the residue 750 cubic centimeters of the wine, to which the compound spirit of orange has been previously added. Finally add the tincture of citro-chloride of iron, syrup, and enough wine to make 1,000 cubic centimeters. Filter if necessary.

II.—For Poultry and Stock.—A good formula for wine of beef and iron is as follows:

Beef extract.....	256 grains
Tincture of iron citro-chloride...	256 minims
Hot water.....	1 fluidounce
Sherry wine enough to make.....	1 pint

Pour the hot water in the beef extract and triturate until a smooth mixture is made. To this add, gradually and under constant stirring, 12 ounces of the wine. Add now, under same conditions, the iron, stir in well, and finally add the remainder of the wine.

Cinchona.—I.—Macerate 100 parts of cinchona succirubra in coarse powder for 30 minutes in 100 parts of boiling water. Strain off the liquor and set aside. Macerate the residuum in 1,000 parts of California Malaga for 24 hours, strain off the liquid and set aside. Finally macerate the magma in 500 parts of alcohol, of 50 per cent, for 1 hour, strain off and set aside. Wash the residue with a little water to recover all the alcoholic tincture; then unite all the

liquids, let stand for 24 hours, and filter. To the filtrate add 800 parts loaf sugar and dissolve by the aid of gentle heat and again filter. The product is all that could be asked of a wine of cinchona. To make a ferrated wine of this, dissolve 1 part of citro-ammoniacal pyrophosphate of iron to every 1,000 parts of wine.

II.—Yvon recommends the following formula:

Red cinchona, coarse powder.....	5 parts
Alcohol, 60 per cent..	10 parts
Diluted hydrochloric acid.....	1 part
Bordeaux wine.....	100 parts

Macerate the bark with the acid and alcohol for 6 days, shaking from time to time, add the wine, macerate for 24 hours, agitating frequently, then filter.

Removal of Musty Taste and Smell from Wine.—For the removal of this unpleasant quality, Kulisch recommends the use of a piece of charcoal of about the size of a hazel nut—5 to 10 parts per 1,000 parts of wine. After this has remained in the cask for 6 to 8 weeks, and during this time has been treated once a week with a chain or with a stirring rod, the wine can be racked off. Obstinate turbidness, as well as stalk taste and pot flavor, can also be obviated by the use of the remedy.

WINTERGREEN, TO DISTINGUISH METHYL SALICYLATE FROM OIL OF.

A quantity of the sample is mixed in a test tube with an equal volume of pure concentrated sulphuric acid. Under these conditions the artificial compound shows no rise in temperature and acquires only a slight yellowish tint, while with the natural oil there is a marked rise in temperature and the mixture assumes a rose-red color, gradually passing into darker shades.

WIRE ROPE.

See also Steel.

A valuable anti-friction and preservative compound for mine cables is as follows: Seven parts soft tallow and 8 parts plumbago, mixed thoroughly; make a long, hollow box or trough, gouge out a 4 by 6 piece of scantling about 2 feet long, sawing it down lengthwise and hollowing out the box or trough enough to hold several pounds of the compound, making also a hole lengthwise of the

trough for the cable to run through; then affix to rope and clamp securely, having the box or trough so fixed that it cannot play, and letting the cable pass through it while going up or down, so that it will get a thorough coating. This, it is found, will preserve a round cable very well, and can be used at least once a week. For a flat steel cable raw linseed oil can be used instead of the tallow, in about the proportion of 6 parts oil and 3 plumbago. If tar is used, linseed oil is to be added to keep the tar from adhering, both ingredients to be mixed while warm.

To preserve wire rope laid under ground, or under water, coat it with a mixture of mineral tar and fresh slaked lime in the proportion of 1 bushel of lime to 1 barrel of tar. The mixture is to be boiled, and the rope saturated with it while hot; sawdust is sometimes added to give the mixture body. Wire rope exposed to the weather is coated with raw linseed oil, or with a paint composed of equal parts of Spanish brown or lamp-black with linseed oil.

WIRE HARDENING:

See Steel.

WITCH-HAZEL JELLY:

See Cosmetics.

Wood

DECORATIVE WOOD-FINISH.

Paint or stencil wood with white-lime paint. When it has dried slowly in the shade, brush it off and a handsome dark-brown tone will be imparted to the oak-wood. Some portions which may be desired darker and redder are stained again with lime, whereby these places become deeper. It is essential that the lime be applied in even thickness and dried slowly, for only then the staining will be red and uniform.

After the staining saturate the wood with a mixture of varnish, 2 parts; oil of turpentine, 1 part; turpentine, $\frac{1}{2}$ part. When the oil ground is dry apply 2 coatings of pale amber varnish.

Colored decorations on pinewood can be produced as follows:

The most difficult part of the work is to remove the rosin accumulations without causing a spot to appear. Burn out the places carefully with a red-hot iron. Great care is necessary to prevent the iron from setting the rosin on fire, thus causing black smoke clouds.

The resulting holes are filled up with plaster to which a little light ochre is added to imitate the shade of the wood as perfectly as possible. Plaster up no more than is necessary.

Rub the wood down with very fine sandpaper, taking especial care to rub only with the grain of the wood, since all cross scratches will remain permanently visible.

After this preliminary work cover the wood with a solution of white shellac, in order not to injure the handsome golden portions of the wood and to preserve the pure light tone of the wood in general.

On this shellac ground paint and stencil with glazing colors, ground with isinglass solution. The smaller, more delicate portions, such as flowers and figures, are simply worked out in wash style with water colors, using the tone of the wood to remain as high lights, surrounding the whole with a black contour.

After this treatment the panels and decorated parts are twice varnished with dammar varnish. The friezes and pilaster strips are glazed darker and set off with stripes; to varnish them use amber varnish.

The style just mentioned does not exclude any other. Thus, for instance, a very good effect is produced by decorating the panels only with a black covering color or with black and transparent red (burnt sienna and a little carmine) after the fashion of *boule* work in rich ornaments, in such a way that the natural wood forms the main part and yet quite a considerable portion of the ornament.

Intarsia imitation is likewise well adapted, since the use of variegated covering colors is in perfect keeping with the decoration of natural wood. How it should be applied, and how much of it, depends upon one's taste, as well as the purpose and kind of the object.

It is a well-known fact that the large pores of oak always look rather smeary, according to whether the workshop is more or less dusty. If this is to be avoided, which is essential for neat work, take good wheat starch, pound it fine with a hammer and stir by means of a wooden spatula good strong polish with the wheat starch to a paste and work the paste into the pores by passing it crosswise over the wood. After about $\frac{1}{2}$ hour, rub down the wood thus treated in such a manner that the pores are filled. In case any open pores remain, repeat the process as before. After that, rub down, polish or deaden. If this operation is not performed, the pores will always look somewhat dirty, despite all

care. Every cabinetmaker will readily perceive that this filling of the pores will save both time and polish in the subsequent finishing.

WOOD FILLERS.

The novice in coach painting is quite as likely to get bewildered as to be aided by much of the information given about roughstuff, the more so as the methods differ so widely. One authority tells us to use a large proportion of lead ground in oil with the coarser pigment, while another says use dry lead and but a small percentage, and still another insists that lead must be tabooed altogether. There are withal a good many moss-grown superstitions associated with the subject. Not the least of these is the remarkably absorbent nature which the surface that has been roughstuffed and "scoured" is supposed to possess. By many this power of absorption is believed to be equal to swallowing up, not only all the color applied, but at least 3 coats of varnish, and none of these would think of applying a coat of color to a roughstuffed surface without first giving it a coat of liquid filler as a sort of sacrificial oblation in recognition of this absorbing propensity. Another authority on the subject has laid down the rule that in the process of scouring, the block of pumice stone must always be moved in one direction, presumably for the reason that some trace of the stone is likely to be visible after the surface is finished.

If the block of stone is scratching, perhaps the appearance of the finished panel may be less objectionable with the furrows in parallel lines than in what engravers call "cross-hatching," but if the rubbing is properly done it is not easy to discover what difference it could make whether the stone is moved in a straight line or a circle. As to absorption, it cannot be distinguished in the finished panel between the surface that was coated with liquid filler and that to which the color was applied directly, except that cracking always occurs much sooner in the former, and this will be found to be the case with surfaces that have been coated with liquid filler and finished without roughstuff. Among the pigments that may be used for roughstuff, and there are half a dozen or more, any of which may be used with success, there is no doubt but that known as "English filler" is best, but it is not always to be had without delay and inconveniences.

Yellow ochre, Reno umber and Key-

stone filler are all suitable for roughstuff, the other having been used many years for the purpose, but, as already remarked, the English filler is best. This is the rule for mixing given by Nobles and Hoare: Four pounds filler, 1 pound ground white lead, 1 pint gold size, 1 pint varnish and $1\frac{1}{2}$ pints turpentine, or $\frac{3}{4}$ pint good size and $\frac{1}{2}$ pint boiled oil in lieu of the varnish. In regard to the use of white lead ground in oil, it makes the rubbing more laborious, increases the liability to scratching, and requires a much longer time to harden before the scouring can be done, without in any appreciable manner improving the quality of the surface when finished.

It may be remarked here that the addition of white lead, whether ground in oil or added dry to the coarser pigment, increases the labor of scouring just in proportion as it is used until sufficient may be used to render the scouring process impossible; hence, it follows that the mixing should be governed by the character of the job in hand. If the job is of a cheap class the use of very little or no lead at all is advisable, and the proportion of Japan and turpentine may also be increased, with the result that a fairly good surface may be obtained with much less labor than in the formula given.

The number of coats of filler required to effect the purpose in any given case must depend upon how well the builder has done his part of the work. If he has left the surface very uneven it follows, as a matter of course, that more coats will be required to make it level, and more of the roughstuff will remain after the leveling process than if the woodwork had been more perfectly done. While the merits of a system or method are not to be judged by its antiquity, there should be a good reason to justify the substitution of a new method for one that has given perfect satisfaction for generations and been used by the best coach painters who ever handled a brush.

A well-known writer on paints says that the effect of a varnish is usually attributed to the manner of its application and the quantity of thinners used for diluting the melted gums, with the prepared oils and the oxidizing agents used in its manufacture. While this has undoubtedly much to do with the successful application of varnish, there are other facts in this connection that should not be overlooked. For example, varnish is sometimes acted on by the breaking up, or the disintegration of the filling coats; which in turn is evidently acted on by the wood itself, according to its nature.

With the aid of the microscope in examining the component parts of wood a cellular tissue is observed which varies in form according to the species and the parts which are inspected. This cellular tissue is made up of small cavities called pores or cells, which are filled with a widely diversified matter and are covered with a hard and usually brittle substance called *lignin*.

This diversified matter consists of mineral salts and various organic substances, gelatinous in their nature and held in solution by a viscous liquid and containing nitrogenous matter in different combinations, the whole being designated by the general name of albuminous substances. The older the wood the more viscous is the matter; while wood of recent growth (sapwood) contains less viscous matter holding these substances in solution. This albumen in wood acts on substances like filler and varnish in one way or the other, good or bad. The seasoning of wood does not dispose of these substances. The water evaporates, leaving them adhering to the sides of the cells. The drier these substances are the less action they exert on the filler or whatever substance is coated on the surface. If the filler disintegrates, it affects the varnish.

All albuminous substances, be they dry or in liquid form, are subject, more or less, according to the protein they contain—which seems, or rather is, the essential principle of all albuminous matter—to the influence of caustic potash and soda. Thus, the albumen of an egg is exactly like that contained in the composition of wood. As albumen in wood becomes solid by drying, it is easily dissolved again, and will then be acted on chemically by any extraneous substance with which it comes in contact.

Some of the shellacs, substitutes for shellacs, and some of the liquid fillers are manufactured from some of the following substances: Old linseed oil, old varnish, old and hard driers, turpentine, benzine, often gasoline, rosin, whiting, cornstarch flour, hulls, paint skins, silica, and so on. The list is long. To these must be added a large volume of potash, to bring it to and hold it in solution. There must be an excess of potash which is not combined into a chemical compound, which if it did, might mitigate its influence on the albumen of the wood. But as there is potash in its pure state remaining in the solution it necessarily attacks the albumen of the wood, causing disintegration, which releases it from the wood, causing white, grayish flakes, and the

formation of a powder. This is not a conclusion drawn from an inference but an established scientific fact resulting from experiments with fillers the various compositions of which were known. All alkalies act on albumen. No one would knowingly varnish over a surface such as it would be were the white of an egg applied to it and then washed with an alkali solution; but that is just what is done when varnish is put over a wood surface filled with a filler which contains an alkali.

Most of the combinations of material used in the painting trade are mixtures; that is, each part remains the same—exerting the same chemical action on another substance, or any other substance coming in contact with a paint mixture will exert the same chemical action on any part, or on any ingredient it contains, the same as if that part was by itself.

We can now account for some of the numerous peculiarities of varnish. We know that any alkali when coming in contact with albumen forms a compound, which on drying is a white, brittle substance easily disintegrated. This is why potash, sal soda, and kindred substances will remove paint. The alkali attacks the albumen in the oil, softening it, causing easy removal, whereas if it were allowed to dry, the albumen in the oil would take on a grayish color quite brittle. Potash or other alkalies in filler not only attack the albumen in the wood, but also attack the albumen in the oil by forming a compound with it. Probably this compound is very slight, only forming a compound in part, enough, nevertheless, to start a destroying influence, which is demonstrated by the following results of experiments. The reader has, perhaps, some time in his career applied a rosin varnish over a potash filler and has been surprised by the good results, a more permanent effect being obtained than in other instances where the best of varnish was used. This is accounted for by the rosin of the potash. Again, the reader may have had occasion to remove varnish with potash and found that potash would not touch it. This is because of its being a rosin varnish. Potash in filler may be rendered somewhat inert, by reason of its compounding with other parts of the filler, but owing to the quantity used in some of the commercial fillers it is not possible that all the alkali is rendered inert. Hence it will attack the albumen wherever found, as all albumen is identical in its chemical composition.

Alkalies have but little effect on the

higher classes of gums, because of their effect on the albumen in the wood and oil. All alcohol varnishes or varnishes made by the aid of heat stand well over an alkali filler. Varnishes which contain a little oil seem to stand well. This is accounted for by the fact that alcohol renders albumen insoluble. Alkalies of all kinds readily attack shellac and several other of the cheap gums, forming unstable compounds on which oil has but little effect.

Close-grained wood contains less albumen and more lignin than open-grained varieties, and consequently does not take so much filler, which accounts for the finish invariably lasting longer than the same kind used on an open-grained wood. Open-grained wood contains more sap than close grained; consequently there is more albumen to adhere to the sides of the cells. The more albumen, the more readily it is attacked by the potash, and the more readily decomposed, or rather destroyed.

Alcohol renders albumen insoluble immediately on application. It prevents it from compounding with any other substance, or any other substance compounding with it. Hence, we must conclude that an application of alcohol to wood before the filler is applied is valuable, which is proven to be a fact by experiment. Wash one half of a board with alcohol, then apply the potash filler over all. Again, wash the portion of the board on which is the filler and apply a heavy-bodied oil varnish. Expose to sunlight and air the same as a finished door or the like, and wait for the result. At the end of a few months a vast difference will be found in the two parts of the surface. The one on which there is no alcohol will show the ravages of time and the elements much sooner than the one on which it is.

Wood finishers demand a difference in the composition of fillers, paste and liquid, for open- and close-grained wood, respectively; but unfortunately they do not demand a difference between either kind in themselves, according to the kind of wood. Paste fillers are used indiscriminately for open-grained wood and liquid for close-grained wood.

To find the fillers best adapted for a certain wood, and to classify them in this respect will require a large amount of chemical work and practical experiments; but that it should be done is evidenced by the fact that both success and failure result from the use of the same filler on different varieties of wood. After once being classified (owing to the large

number now on the market), they will not number nearly so many in the aggregate as might be supposed; as it will be found in many instances that two entirely different varieties of wood resemble each other more closely in their vascular formation and cell characteristics than do two other specimens of the same variety. It is a recognized fact that paste fillers whose base is starch or the like work better and give better results in certain instances, while those whose base is mineral matter seem to do better in other cases.

It is noticed that rosewood as a finishing veneer is obsolete. This is not because of its scarcity, but because it is so hard to finish without having been seasoned for a long time. In these days, manufacturers cannot wait. It takes longer for the sap of rosewood to become inactive, or in trade parlance to "die," than any other wood. This is because it takes so long for the albumen in the sap to coagulate. Rosewood has always been a source of trouble to piano makers, on account of the action of the sap on the varnish. However, if this wood, previously to filling, was washed with a weak solution of phosphoric acid, and then with wood spirit, it might be more easily finished. The phosphoric acid would coagulate the albumen on the surface of the wood immediately, while alcohol would reduce it to an insoluble state. The idea here is to destroy the activity of the sap, on the same principle as sappy places and knot sap are destroyed by alcohol-shellac before being painted.

Oak is another wood which gives the painter trouble to finish. This may be accounted for as follows: Oak contains a sour acid principle called tannic acid. It is a very active property. Wood during the growing season contains more albumen; thus in the circulation of the sap a large quantity of soft matter is deposited on the lignin which lines the cells, which lignin, if it contains any acid matter, acts on the material of the filler. Tannic acid has a deleterious effect on some of the material of which a number of fillers are made. Starch and many gums are susceptible to its influence, making some of them quite soft. Oak, like most other timber cut at the season when the least sap is in circulation, is the more easily finished.

The vascular formation may, and no doubt has, something to do with wood finishing. Different species of wood differ materially in their vascular and cellular formation. Wood finishers recognize a difference in treatment of French burl walnut and the common American

variety. Circassian and Italian walnut, although of the same species, demand widely different treatment in finishing to get the best results.

The only way to find the best materials to use in certain cases is to study and experiment with that end in view. If, by aid of a microscope, a certain piece of wood shows the same cellular formation that another piece did which was successfully finished by a certain process, it may be regarded as safe to treat both alike. If observation on this line is indulged in, it will not take the finisher very long to learn just what treatment is best for the work in hand. How often it has been noticed in something of two parts, like a door, that the panels when finished will pit, run, or sag, while the sides will present a surface in every way desirable and *vice versa*. This is due to the difference in the cellular construction of the wood and to the cellulose, and cannot be otherwise for the parts have been seasoned the same time and treated exactly alike. The physiology of wood is imperfectly understood, but enough is known to warrant us in saying with a certainty that the chemicals in fillers do act upon the principles embodied in its formation.

Some tried formulas follow:

I.—Make a paste to fill the cracks as follows: Old furniture polish: Whiting, plaster of Paris, pumice stone, litharge, equal parts, Japan drier, boiled linseed oil, turpentine, coloring matter, of each a sufficient quantity.

Rub the solids intimately with a mixture of 1 part of the Japan, 2 parts of the linseed oil, and 3 parts of turpentine, coloring to suit with vandyke brown or sienna. Lay the filling on with a brush, let it set for about 20 minutes, and then rub off clean except where it is to remain. In 2 days it will be hard enough to polish. After the surface has been thus prepared, the application of a coat of first-class copal varnish is in order. It is recommended that the varnish be applied in a moderately warm room, as it is injured by becoming chilled in drying. To get the best results in varnishing, some skill and experience are required. The varnish must be kept in an evenly warm temperature, and put on neither too plentifully nor too gingerly. After a satisfactorily smooth and regular surface has been obtained, the polishing proper may be done. This may be accomplished by manual labor and dexterity, or by the application of a very thin, even coat of a very fine, transparent varnish.

If the hand-polishing method be preferred, it may be pursued by rubbing briskly and thoroughly with the following finishing polish:

Alcohol.....	8 ounces
Shellac.....	2 drachms
Gum benzoin.....	2 drachms
Best poppy oil.....	2 drachms

Dissolve the shellac and gum in the alcohol in a warm place, with frequent agitation, and, when cold, add the poppy oil. This may be applied on the end of a cylindrical rubber made by tightly rolling a piece of flannel, which has been torn, not cut, into strips 4 to 6 inches wide. It should be borne in mind that the surface of the cabinet work of a piano is generally veneered, and this being so, necessitates the exercise of much skill and caution in polishing.

II.—Prepare a paste from fine starch flour and a thick solution of brown shellac, with the spatula upon a grinding stone, and rub the wooden object with this. After the drying, rub off with sandpaper and polish lightly with a rag moistened with a thin shellac solution and a few drops of oil. The ground thus prepared varnish once or twice and a fine luster will be obtained. This method is well adapted for any wood with large pores, such as oak.

Removal of Heat Stains from Polished Wood.—Fold a sheet of blotting paper a couple of times (making 4 thicknesses of the paper), cover the place with it, and put a hot smoothing iron thereon. Have ready at hand some bits of flannel, also folded and made quite hot. As soon as the iron has made the surface of the wood quite warm, remove the paper, etc., and go over the spot with a piece of paraffine, rubbing it hard enough to leave a coating of the substance. Now with one of the hot pieces of flannel rub the injured surface. Continue the rubbing, using freshly warmed cloths until the whiteness leaves the varnish or polish. The operation may have to be repeated.

PRESERVATION OF WOOD.

I.—An excellent way of preserving wood is to cut it between August and October. The branches are removed, leaving only the leaves at the top. The trunks, carefully cut or sawn (so that their pores remain open), are immediately placed upright, with the lower part immersed in tanks three-quarters filled with water, into which 3 or 4 kilograms of powdered cupric sulphate per hectoliter have been introduced. The mass of

leaves left at the extremity of each trunk is sufficient to cause the ascent of the liquid by means of the capillary force and a reserve of energy in the sap.

II.—Wood which can be well preserved may be obtained by making a circular incision in the bark of the trees a certain time before cutting them down. The woodcutters employed in the immense teak forests of Siam have adopted in an empirical way a similar process, which has been productive of good results. The tree is bled, making around the trunk, at the height of 4 feet above ground, a circular incision 8 inches wide and 4 inches deep, at the time when it is in bloom and the sap rising. Sometimes the tree is left standing for 3 years after this operation. Frequently, also, a deep incision reaching the heart is made on two opposite sides, and then it takes sometimes only 6 months to extract the sap.

It is probable that it is partly in consequence of this method that the teak-wood acquires its exceptional resistance to various destructive agents.

III.—A good preservation of piles, stakes, and palisades is obtained by leaving the wood in a bath of cupric sulphate of 4° of the ordinary acidimeter for a time which may vary from 8 to 15 days, according to greater or less dryness of the wood and its size. After they are half dried they are immersed in a bath of limewater; this forms with the sulphate an insoluble compound, preventing the rain from dissolving the sulphate which has penetrated the wood. This process is particularly useful for vine props and the wood of white poplars.

A good way to prevent the decay of stakes would be to plant them upside down; that is, to bury the upper extremity of the branch in the ground. In this way, the capillary tubes do not so easily absorb the moisture which is the cause of decay. It frequently happens that for one or another reason, the impregnation of woods designed to be planted in the ground, such as masts, posts, and supports has been neglected. It would be impracticable, after they are placed, to take up these pieces in order to coat them with carbolineum or tar, especially if they are fixed in a wall, masonry, or other structure. Recourse must be had to other means. Near the point where the piece rises from the ground, a hole about one centimeter in width is made in a downward slanting direction, filled with carbolineum, and closed with a wooden plug.

It depends upon the consistency of the wood whether the liquid will be absorbed in 1 or 2 days. The hole is filled again for a week. The carbolineum replaces by degrees the water contained in the wood. When it is well impregnated, the hole is definitely closed with a plug of wood, which is sawn level with the opening. The wood will thus be preserved quite as well as if it had been previously coated with carbolineum.

IV.—Wooden objects remaining in the open air may be effectually protected against the inclemency of the weather by means of the following coating: Finely powdered zinc oxide is worked into a paste with water and serves for whitewashing walls, garden fences, benches, and other wooden objects. After drying, probably at the end of 2 or 3 hours, the objects must be whitewashed again with a very dilute solution of zinc chloride in glue or water. Zinc oxide and zinc chloride form a brilliant, solid compound, which resists the inclemency of the weather.

As a paint for boards, planks for covering greenhouses, garden-frames, etc., Inspector Lucas, of Reutlingen (Württemberg), has recommended the following coating: Take fresh cement of the best quality, which has been kept in a cool place, work it up with milk on a stone until it is of the consistency of oil paint. The wood designed to receive it must not be smooth, but left rough after sawing. Two or 3 coats are also a protection from fire. Wood to be thus treated must be very dry.

V.—Wood treated with creosote resists the attacks of marine animals, such as the teredo. Elm, beech, and fir absorb creosote very readily, provided the wood is sound and dry. Beechwood absorbs it the best. In fir the penetration is complete, when the wood is of a species of rapid growth, and of rather compact grain. Besides, with the aid of pressure it is always possible to force the creosote into the wood. Pieces of wood treated with creosote have resisted for 10 or 11 years under conditions in which oak wood not treated in this way would have been completely destroyed.

The prepared wood must remain in store at least 6 months before use. The creosote becomes denser during this time and causes a greater cohesion in the fibers. In certain woods, as pitch pine, the injection is impossible, even under pressure, on account of the presence of rosin in the capillary vessels.

VI.—M. Zironi advises heating the wood

in vacuo. The sap is eliminated in this way. Then the receiver is filled with rosin in solution with a hydrocarbide. The saturation takes place in two hours, when the liquid is allowed to run off, and a jet of vapor is introduced, which carries off the solvent, while the rosin remains in the pores of the wood, increasing its weight considerably.

VII.—Wood can be well preserved by impregnating it with a solution of tannate of ferric protoxide. This method is due to Hazfeld.

VIII.—The Hasselmann process (xylolized wood), which consists in immersing the wood in a saline solution kept boiling under moderate pressure, the liquid containing copper and iron sulphates (20 per cent of the first and 80 per cent of the second), as well as aluminum and kainit, a substance until recently used only as a fertilizer, is now much employed on the railways in Germany.

IX.—Recently the discovery has been made that wood may be preserved with dissolved betuline, a vegetable product of the consistency of paste, called also birchwood rosin. Betuline must first be dissolved. It is procurable in the crude state at a low price. The wood is immersed for about 12 hours in the solution, at a temperature of from 57° to 60° F.

After the first bath the wood is plunged into a second, formed of a solution of pectic acid of 40° to 45° Bé., and with a certain percentage of an alkaline carbonate—for instance, potassium carbonate of commerce—in the proportion of 1 part of carbonate to about 4 parts of the solution. The wood remains immersed in this composition for 12 hours; then it is taken out and drained from 8 to 15 hours, the time varying according to the nature of the wood and the temperature. In consequence of this second bath, the betulin which was introduced through the first immersion, is fixed in the interior of the mass. If it is desirable to make the wood more durable and to give it special qualities of density, hardness, and elasticity, it must be submitted to strong pressure. In thus supplementing the chemical with mechanical treatment, the best results are obtained.

X.—A receiver of any form or dimensions is filled with a fluid whose boiling point is above 212° F., such as heavy tar oil, saline solutions, etc. This is kept at an intermediate temperature varying between 212° F. and the boiling point; the

latter will not be reached, but if into this liquid a piece of wood is plunged, an agitation analogous to boiling is manifested, produced by the water and sap contained in the pores of the wood. These, under the action of a temperature above 212° F., are dissolved into vapor and traverse the bath.

If the wood is left immersed and a constant temperature maintained until every trace of agitation has disappeared, the water in the pores of the wood will be expelled, with the exception of a slight quantity, which, being in the form of vapor, represents only the seventeen-hundredth part of the original weight of the water contained; the air which was present in the pores having been likewise expelled.

If the liquid is left to cool, this vapor is condensed, forming a vacuum, which is immediately filled under the action of the atmospheric pressure. In this way the wood is completely saturated by the contents of the bath, whatever may be its form, proportions or condensation.

To attain the desired effect it is not necessary to employ heavy oils. The latter have, however, the advantage of leaving on the surface of the prepared pieces a kind of varnish, which contributes to protect them against mold, worms, moisture, and dry rot. The same phenomenon of penetration is produced when, without letting the wood grow cold in the bath, it is taken out and plunged immediately into a cold bath of the same or of a different fluid. This point is important, because it is possible to employ as fluids to be absorbed matters having a boiling point below 212° F., and differing in this respect from the first bath, which must be composed of a liquid having a boiling point above 212° F.

If, instead of a cold bath of a homogeneous nature, two liquids of different density separated in two layers, are employed, the wood can, with necessary precautions, be immersed successively in them, so that it can be penetrated with given quantities of each. Such liquids are heavy tar oil and a solution of zinc chloride of 2° to 4° Bé. The first, which is denser, remains at the bottom of the vessel, and the second above. If the wood is first immersed in a saline solution, it penetrates deep into the pores, and when finally the heavy oil is absorbed, the latter forms a superficial layer, which prevents the washing out of the saline solution in the interior, as well as the penetration of moisture from the outside.